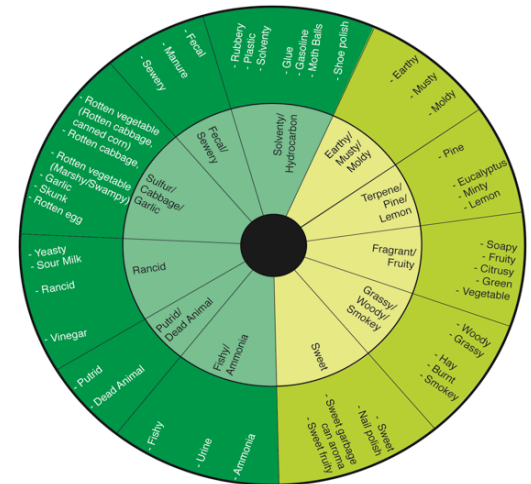


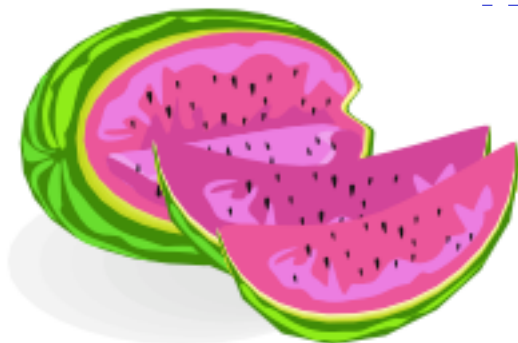
Odor Wheel Applications

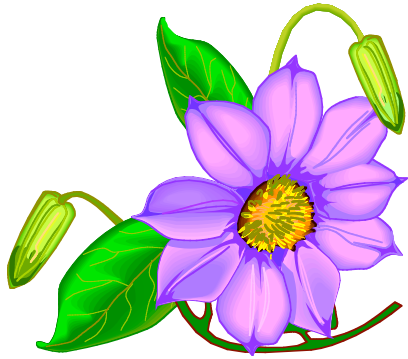


Mel Suffet

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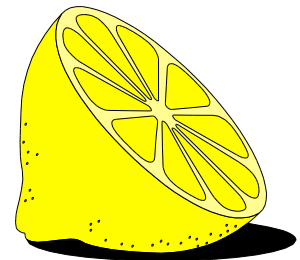
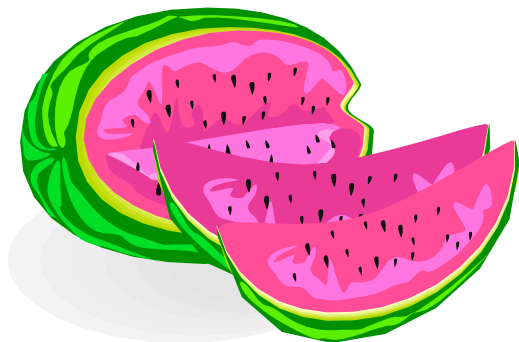


Character of Odor

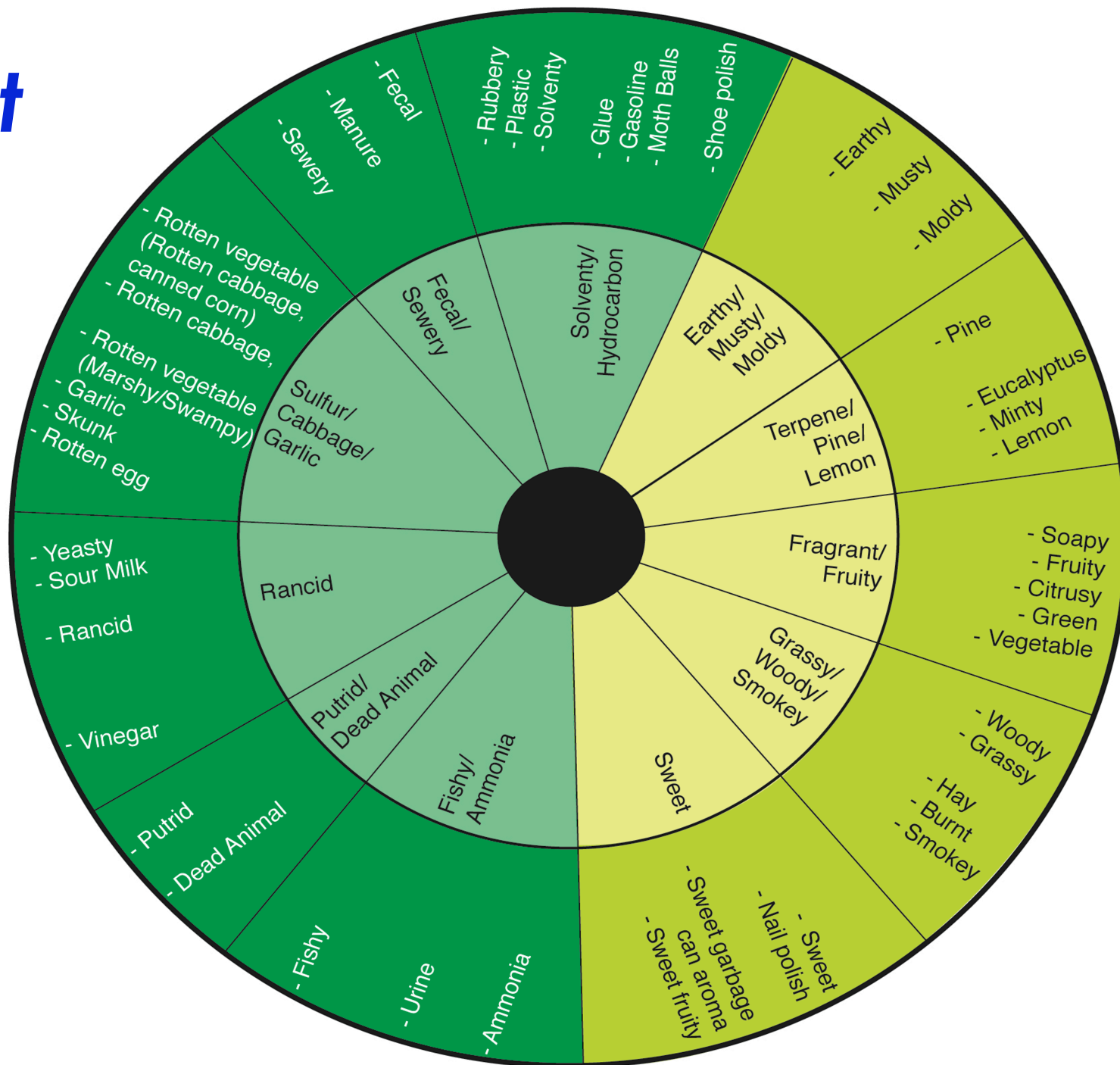


Naming the Odor,
so that Everyone
Understands the Problem.

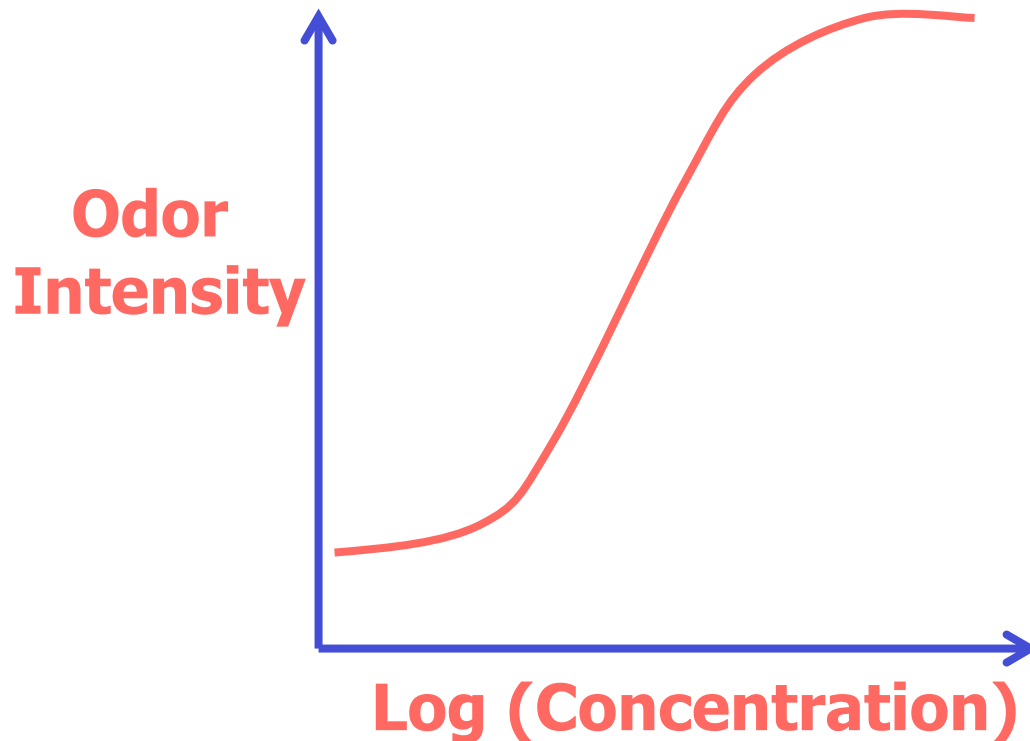
- Critical to Understanding Odor Nuisance
- Naming Consensus Between Panelists
and the Public



Compost Odor Wheel



EACH Odorant Concentration versus Odor Intensity



$$\text{Odor Intensity} = m \log [\text{conc.}] + b$$

Weber-Fechner Law

$$\log (\text{Odor Intensity}) = m \log [\text{conc.}] + b$$

Steven's Power Law

- Semi-logarithmic relationship
- Different chemicals have different curves

RELEASING ODORANTS INTO AIR



Case Study 1. Compost Treatment of Raw Sludge from a WWTP

Sludge A	Avg.	S.D.
Hay/Manure	3.5	3
Sewery/Fecal	2.5	2.2
Ammonia	1.5	0.9
Earthy	1.5	1.7
Rotten vegetable	1.3	1.6
Burnt	note	

Sludge B	Avg.	S.D.
Ammonia	7.5	0.9
Earthy	2	2
Rotten fishy	1.5	1.7
Dead animal	note	
Hay	note	

Sludge C	Avg.	S.D.
Fecal	5.5	0.9
Hay	3	2.2
Rancid	2	2.4
Ammonia	1.3	1.6
Earthy	note	
Rotten vegetable	note	

Raw Sludge Odors

Fecal

Sewery/Fecal

Rotten vegetable

Rancid

Ammonia

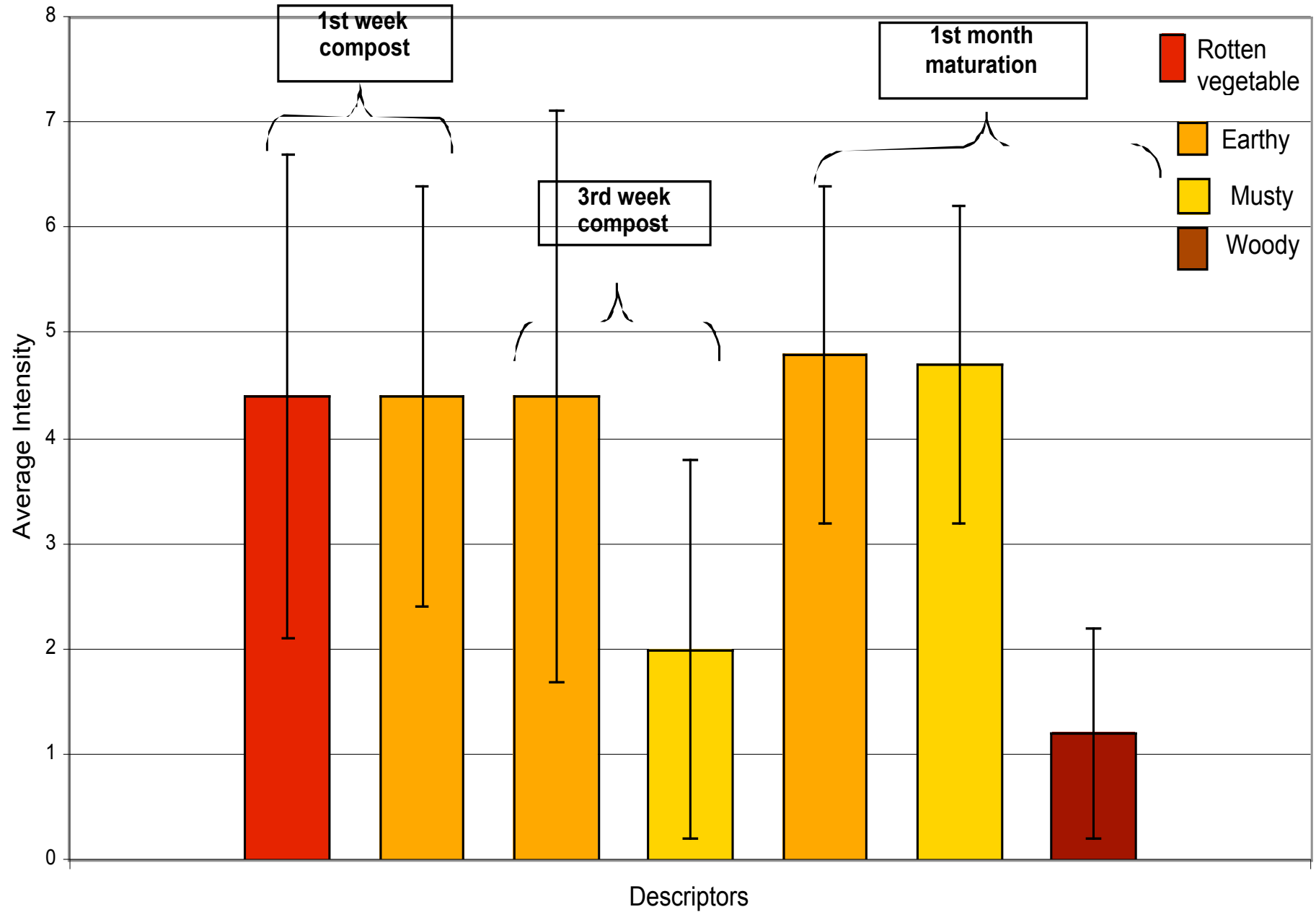
Rotten fishy

Hay/Manure

Hay

Earthy/Musty

Compost from Bury 07/12/2006



Case Study 2. 4/28/06

Biofilter Results

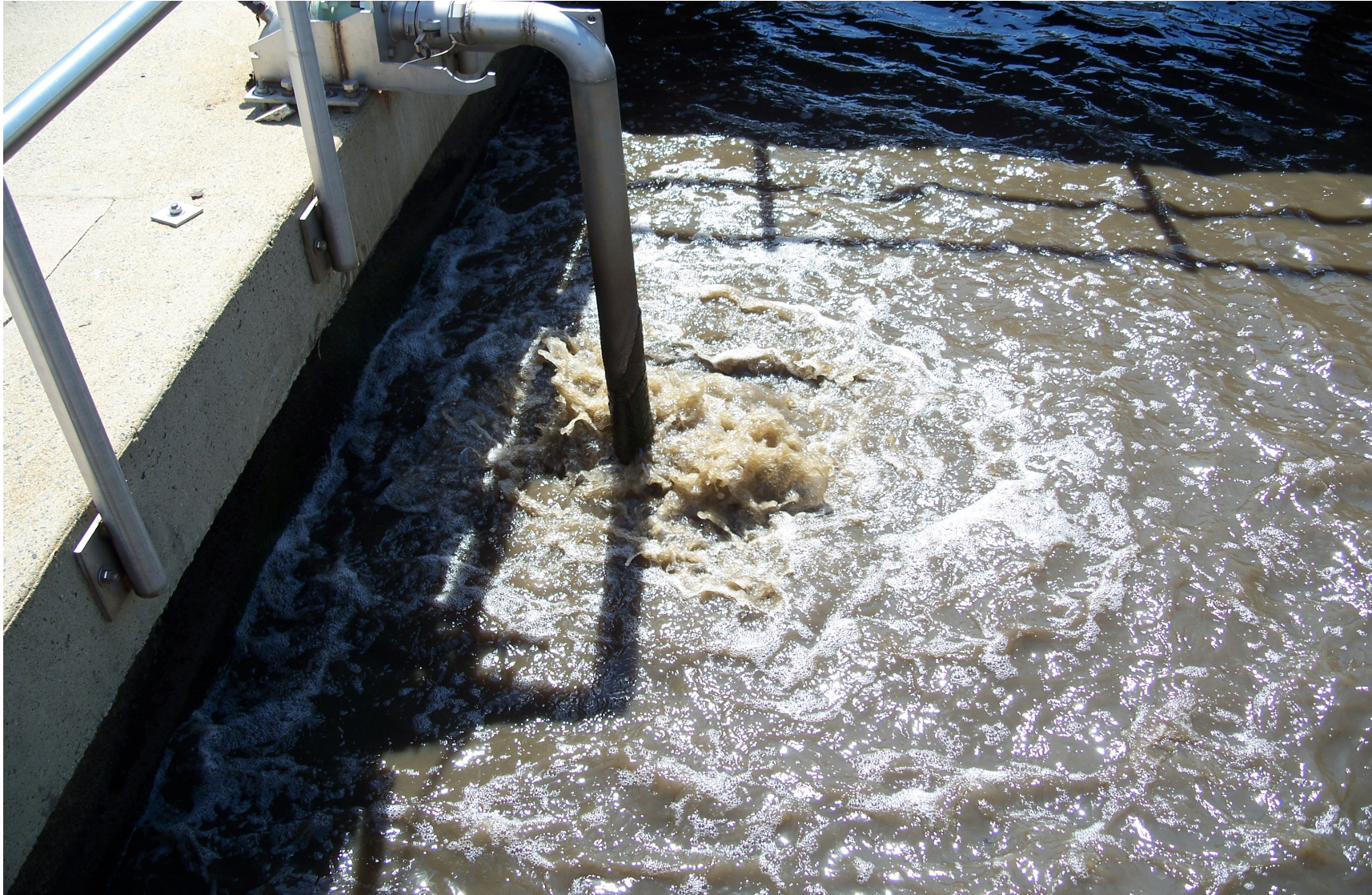
Biofilter In	Avg.	S.D.	Biofilter Out	Avg.	S.D.
Cabbage	6	2.4	Cabbage	3	3
Fish	5	3.3	Fish	1.5	1.7
Sewery	note		Rancid	note	
Acid	note		Fecal	note	
Earthy	note		Oxidant	note	
			Pine	note	

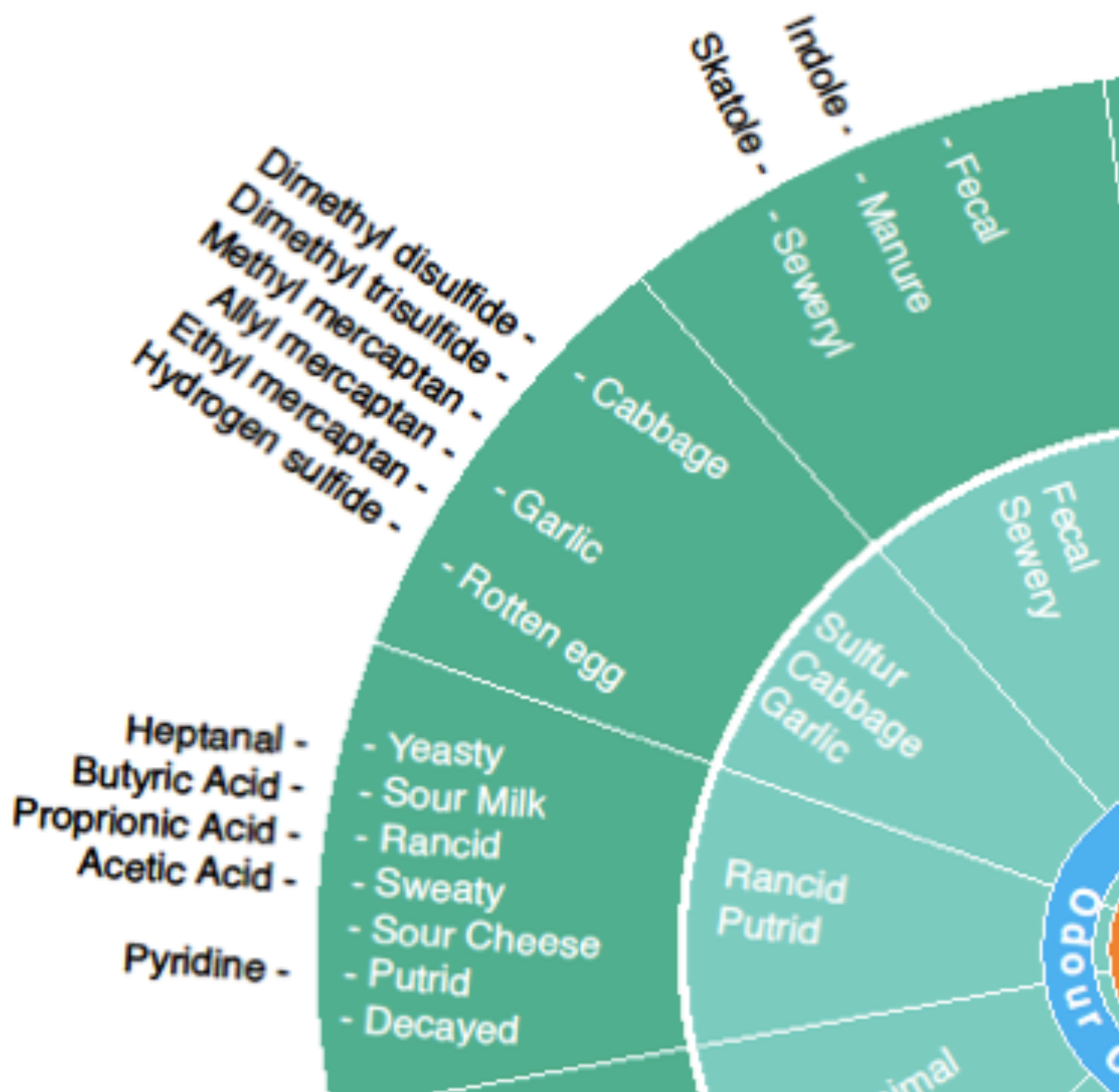
Note: Basis of quantification - H₂S

Case 3. Odors-Biological WW Treatment

<u>Odor Type</u> <u>(60 samples/yr)</u>	<u>Times used in Consensus</u>	
	<u>Year 1</u>	<u>Year 2</u>
oniony	60	38
garlicky	29	18
dec.vegetation	29	39
earthy	20	42
briny	19	2
musty	15	8
solventy	12	14
creeky	11	20
6 others	<10 identifications	

Aerated activated sludge is one source of odors





Case Study 3- 2001-5

- Odor surveys in the community at one wastewater plant had identified a particular persistent odor
- The odor team described it as “canned corn” (rotten cabbage) in odor quality
- The odor team agreed that is smelled just like an odor standard of dimethyl sulfide
- **MICROBIOLOGICALLY:**
 - $\text{R-O-CH}_3 + \text{H}_2\text{S} \rightarrow \text{R-OH} + \text{CH}_3\text{SH}$
 - $\text{R-O-CH}_3 + \text{CH}_3\text{SH} \rightarrow \text{R-OH} + \text{CH}_3\text{SCH}_3$
 - $\text{CH}_3\text{SH} + \text{CH}_3\text{SH} + \text{O}_2 \rightarrow \text{H}_2\text{O} + \text{CH}_3\text{SSCH}_3$

Typical Levels of DMS in 2004/ 2005 at the WWT Plant

Location	Oct – Nov 2003 (avg)	April 2004	Aug – Sept 2004 (3 washes from R&H)	May (avg) 2005 (uniform discharges)
R&H	----	9-63	ND – 706	47
PST #1 In	----	12-176	35 – 5980	457
PST #1 Out	----	16-149	50 – 4800	520
PST #2 Out	----	----	----	487
Aeration Tanks	561 (ppb)	11-215	15 – 1510	72, 164
RAS	----	15-98	74 – 3200	441
Plant Effluent	----	----	----	ND

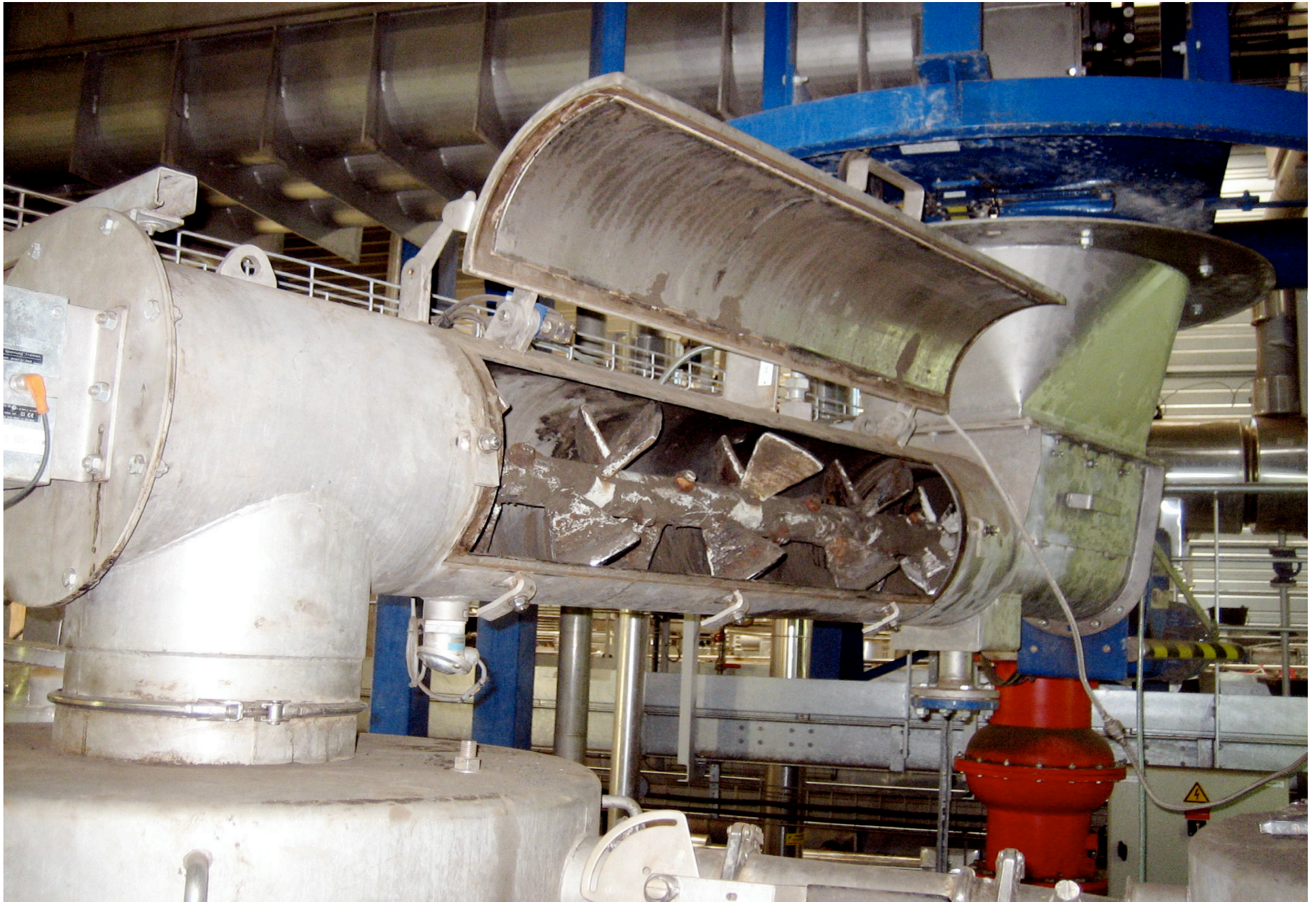
Dimethyl Sulfide (DMS)

- DMS was unusually high compared to methyl mercaptan and dimethyl trisulfide
- An industrial source in the sewer collection system was tracked down
- The industry was discharging dimethyl sulfoxide (DMSO)
- Microbiologically DMSO → DMS

Dimethyl Sulfoxide (DMSO)

- Studies were conducted in cooperation with the industrial source to reduce discharges of the DMSO
- DMSO below levels of 400 µg/L have reduced the DMS to about 50 µg/L or less, and the “canned corn” (rotten cabbage) DMS odor has not been a problem

Case Study 4. Sludge Drying Odors



STEP 1.CLASSIFYING ODOR CATEGORIES

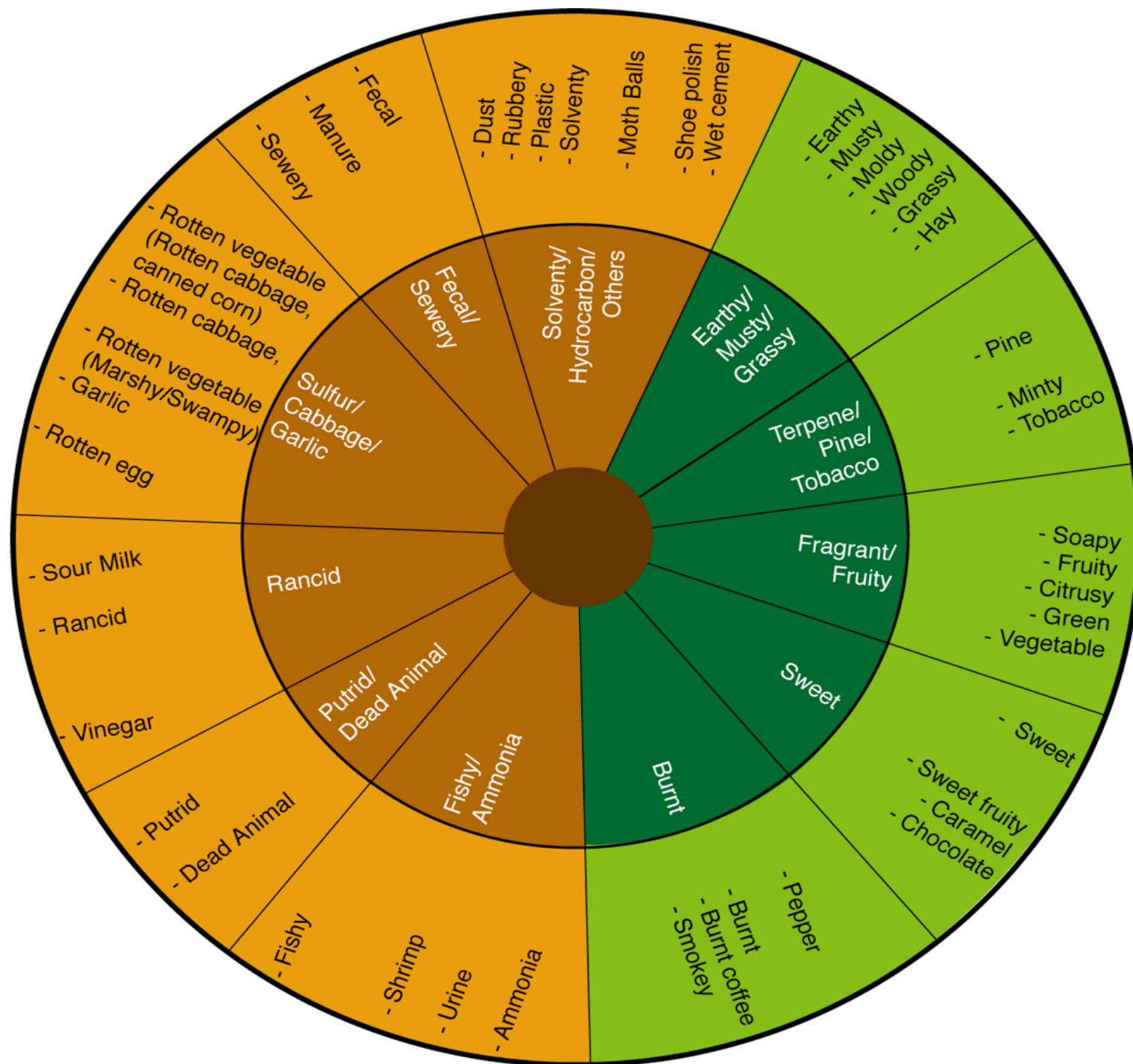
33 dried sludge samples at different stages of treatment plus samples generated by a lab sludge drying pilot

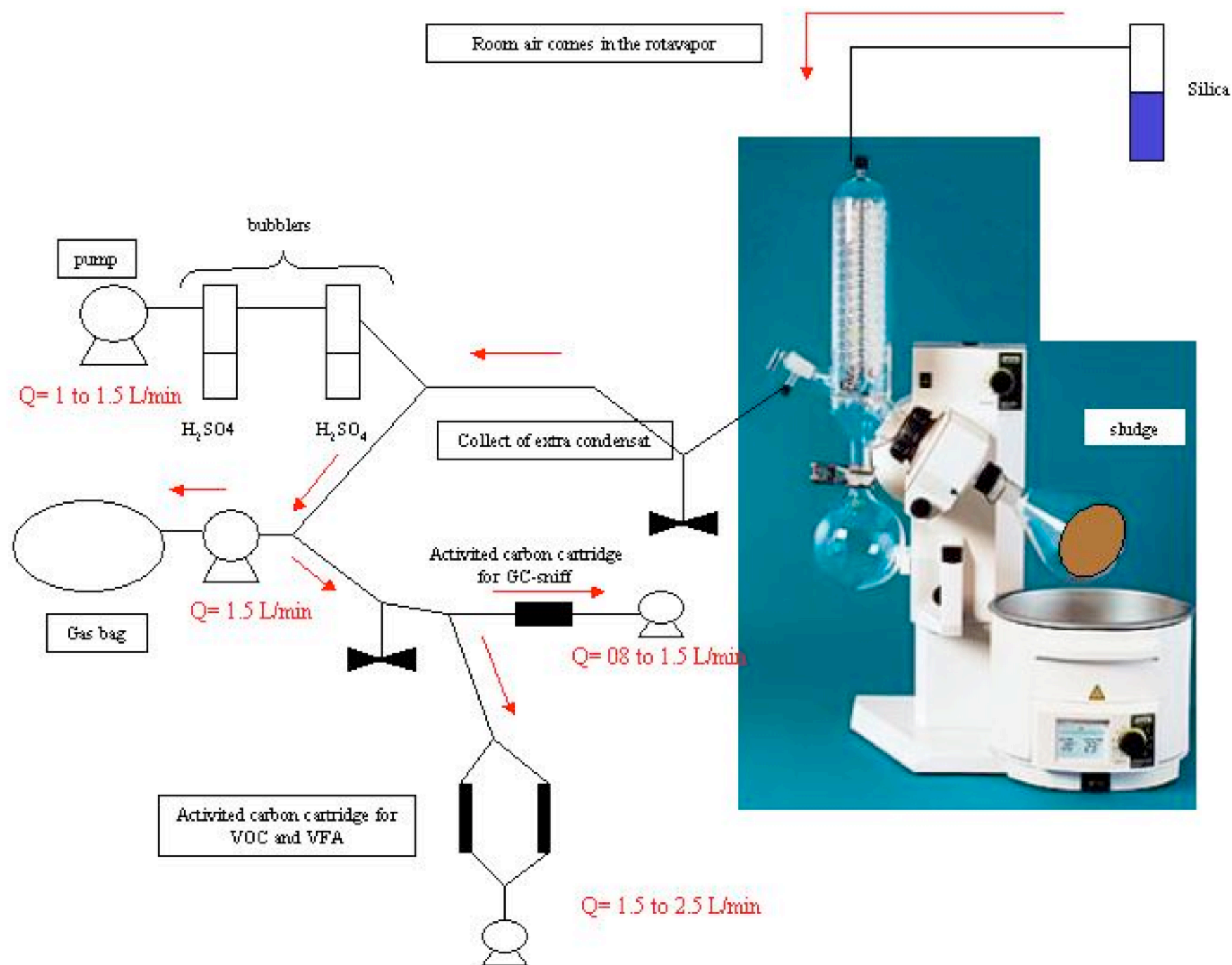
<u>SAMPLE 11</u>		
	Avg.	S.D.
Fecal	4,2	3,6
Earthy	2,9	2,7
Fishy	2,8	3,3
Ammonia	2,4	2,2
Manure	note	
Dead animal	note	
Rancid	note	
Bunt coffee	note	
Rotten cabbage	note	

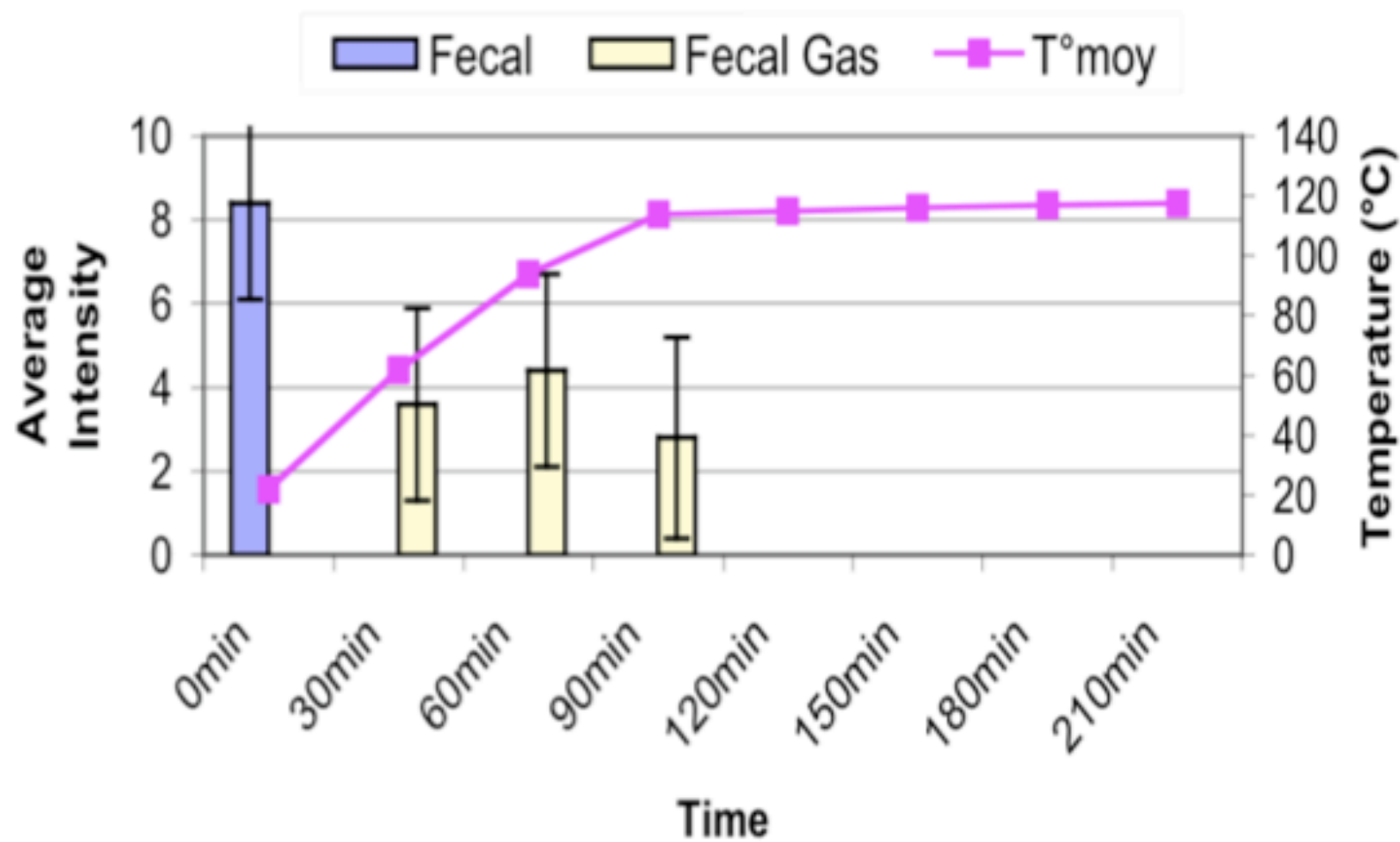
Table 1. Example of a OPM
Results for a Sludge Dried Sample

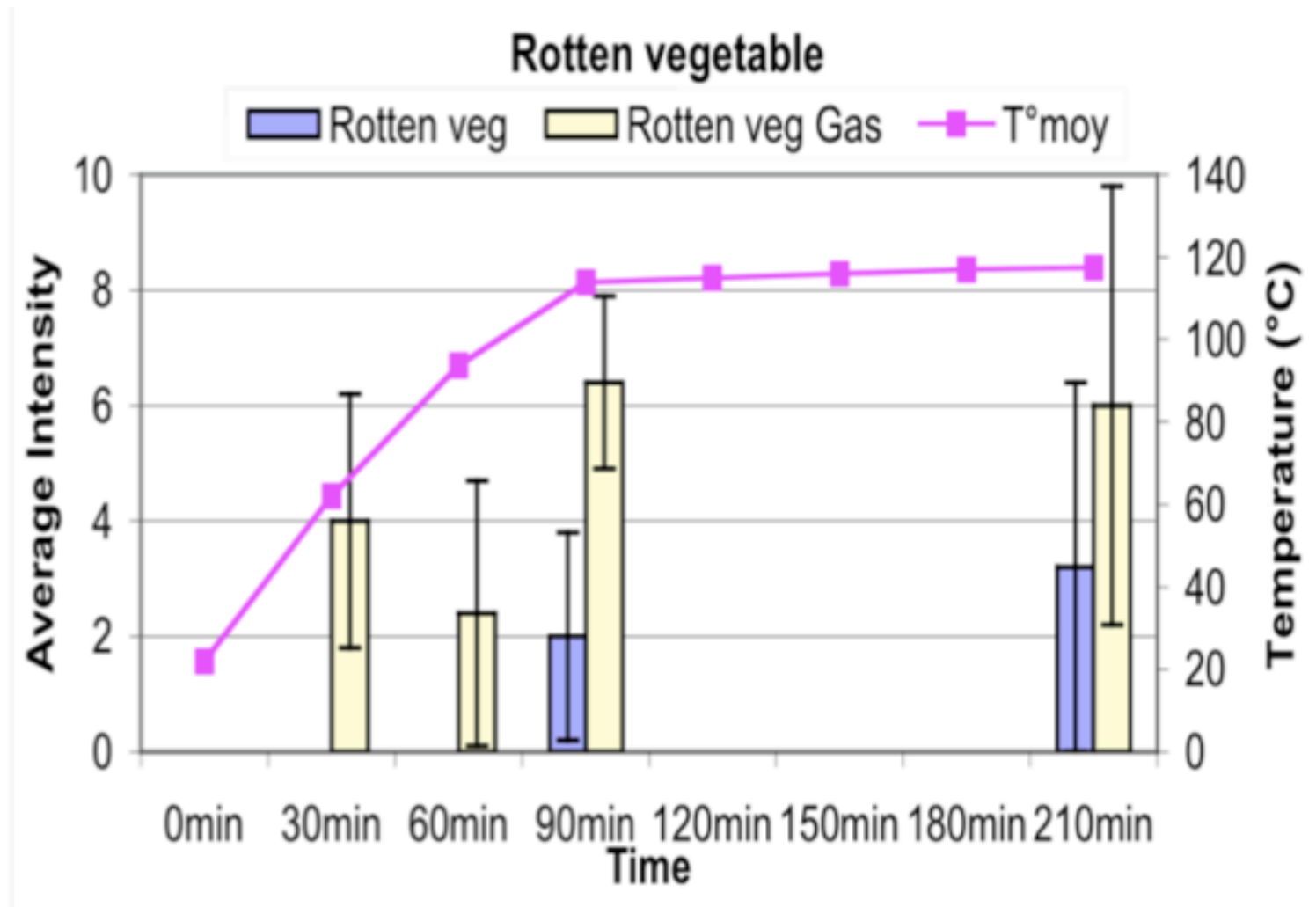
Odor	Total samples	Number of samples with the odor	Intensity Avg. ^a
Rotten fishy	33	30	4.1
Manure/Hay	33	28	3.6
Earthy/Musty	33	29	2.4
Rancid	33	19	2.2
Shrimp	33	8	2.2
Fecal	33	28	2
Sweet	33	3	1.9
Burnt coffee	33	4	1.9
Ammonia	33	21	1.8
Burnt	33	15	1.6
Rotten vegetable	33	5	1.5
Dead Animal	33	5	1.2
Sewery	33	5	1.1

Sludge Drying Odor Wheel

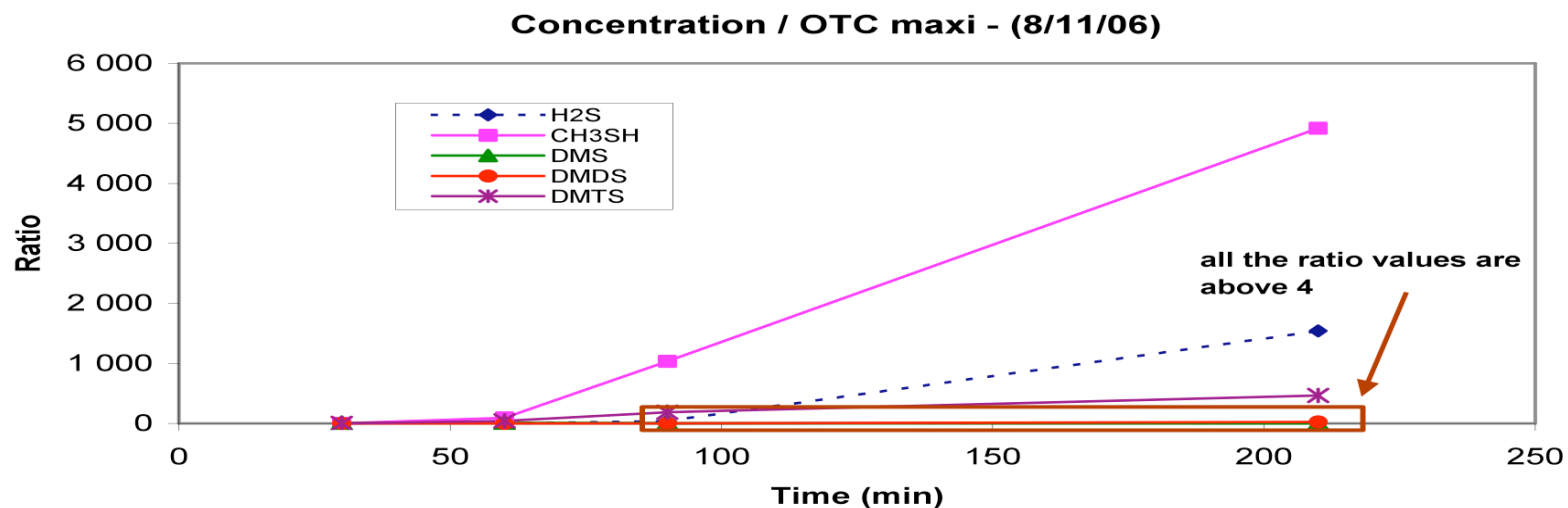
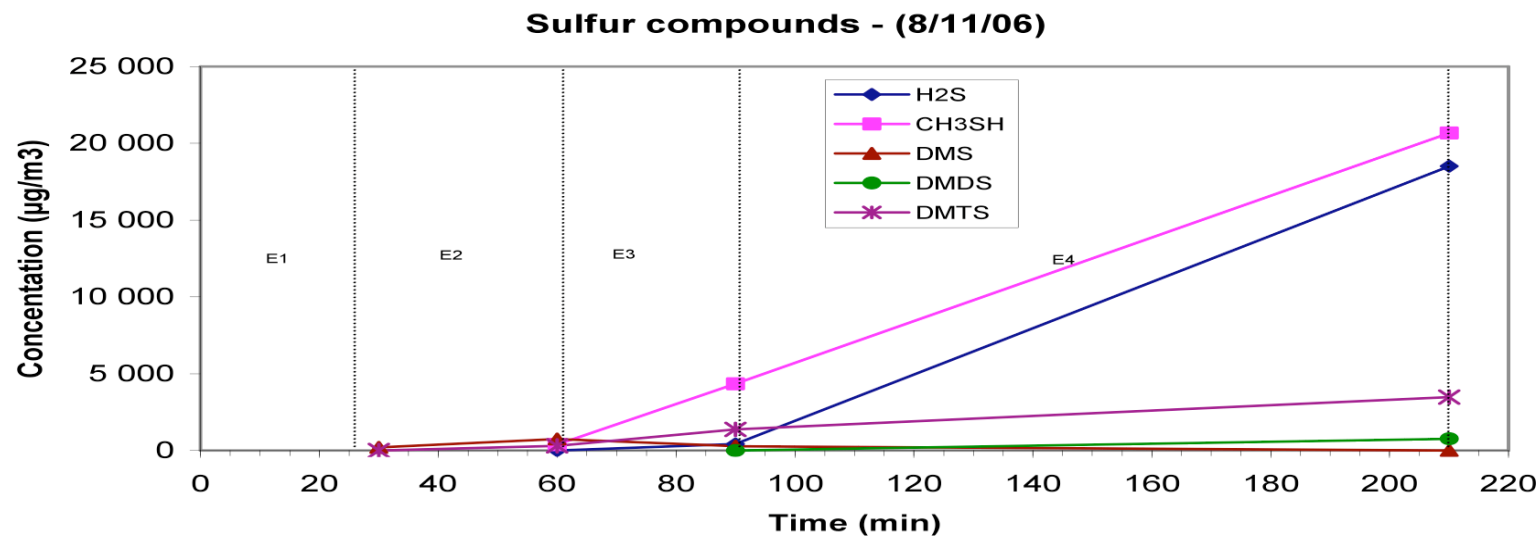




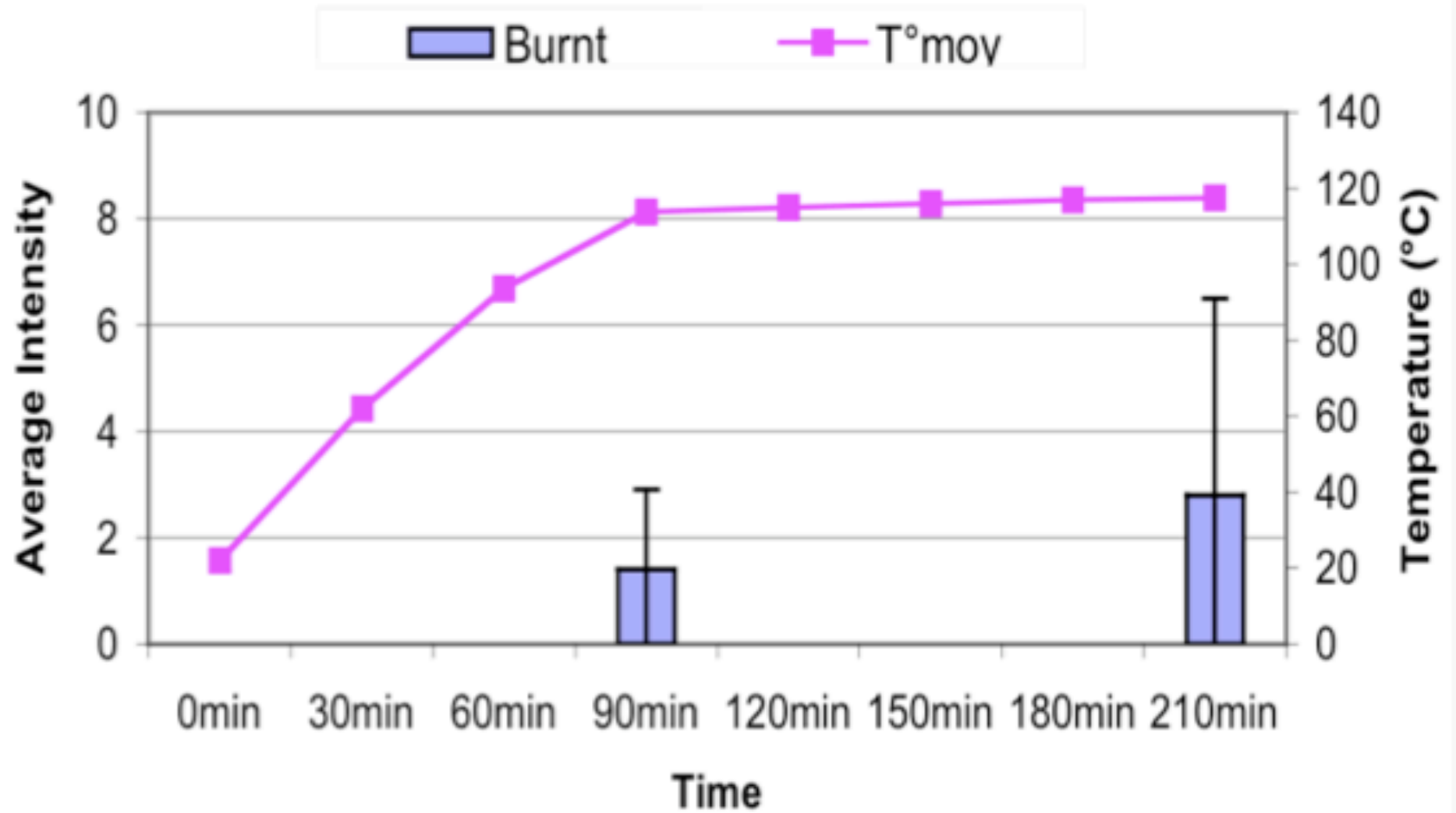




STEP 2. PHYSICO-CHEMICAL ANALYSIS



Burnt

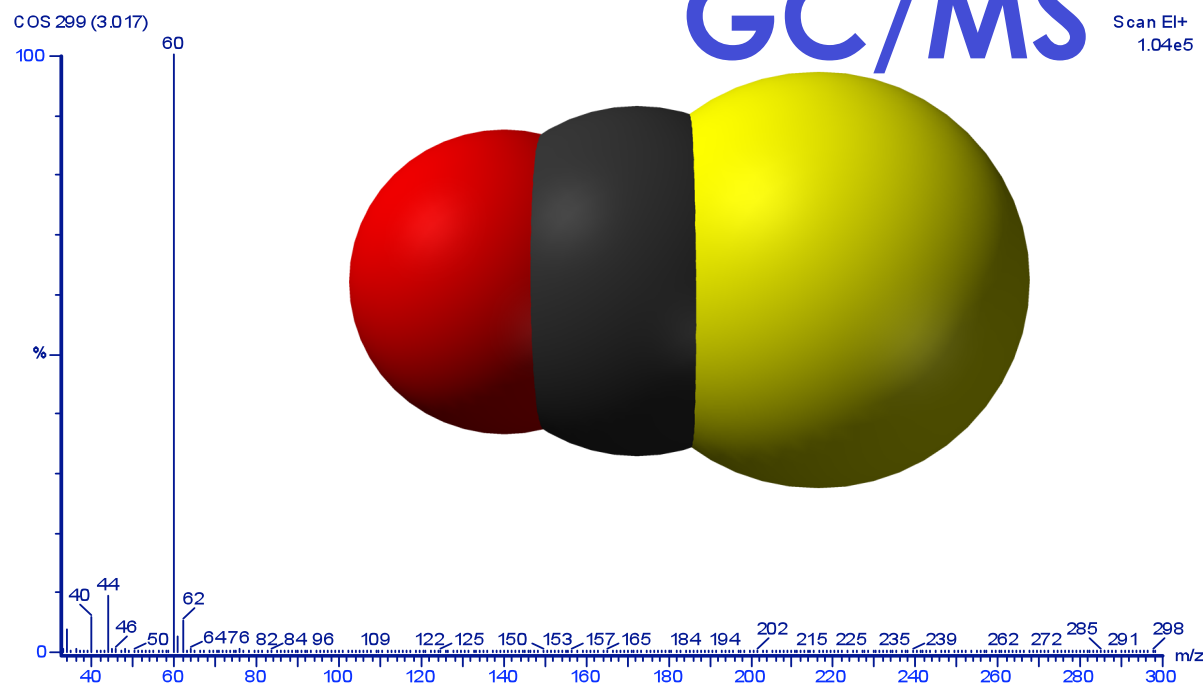


STEP 3. CONFIRMATION OF ODORANTS BY OLFACTORY- GC/MS

Correspondence between odour descriptors and specific chemicals as determined by olfactory-GC/MS (sample 2)

Retention time (min)	Odour at olfactory port	Compound identified
3.37	Burnt-Rotten egg	Carbonyl sulphide (COS)
8.39	Vinegar	Acetic acid
9.38	Rotten cabbage- garlic	Dimethyldisulfide
17.20	Rancid	Butyric acid

STEP 3. CONFIRMATION OF ODORANTS BY OLFACTORY- GC/MS

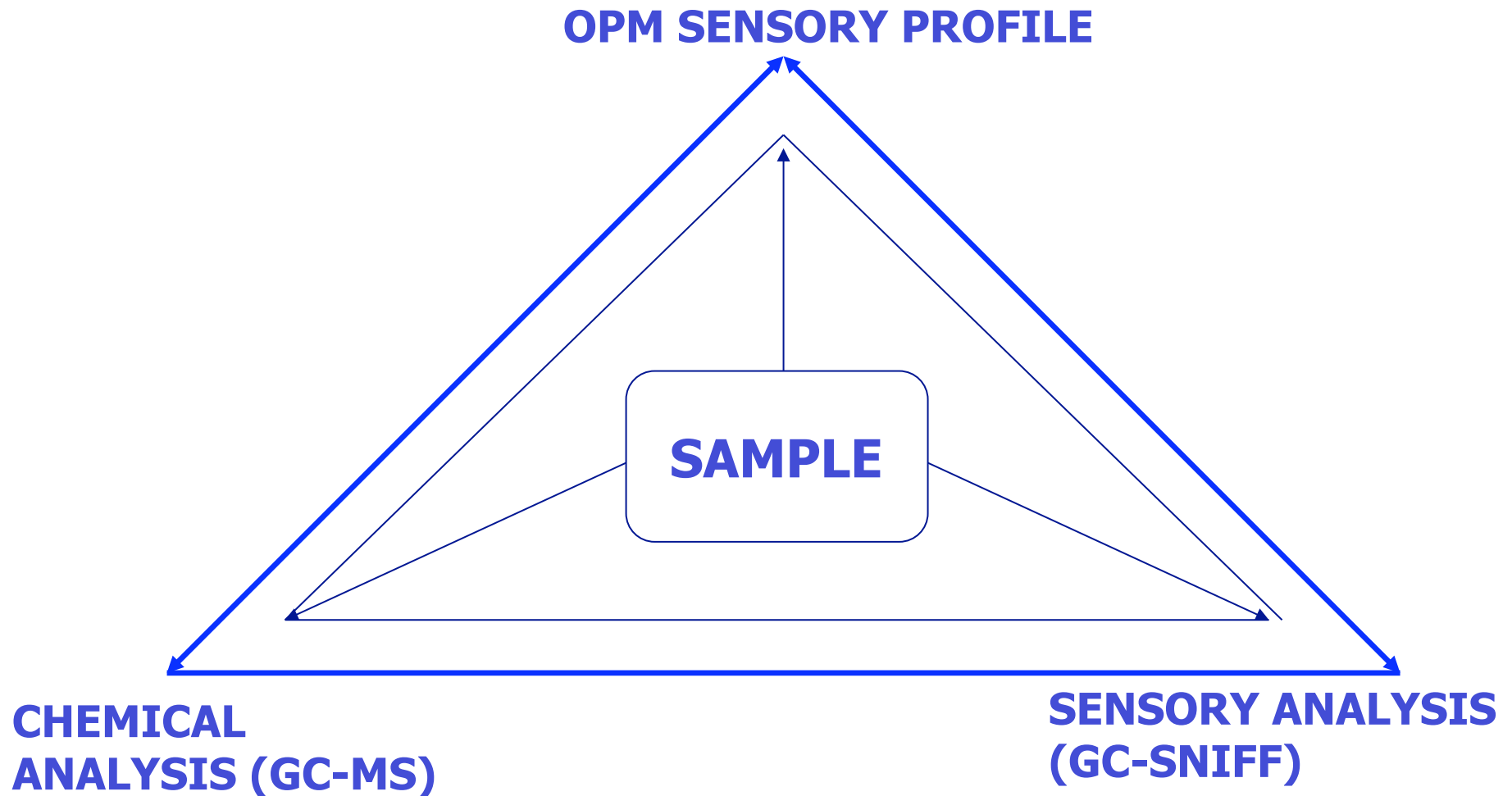


*-Can form
from the
reaction of
sulfur with
C=O*

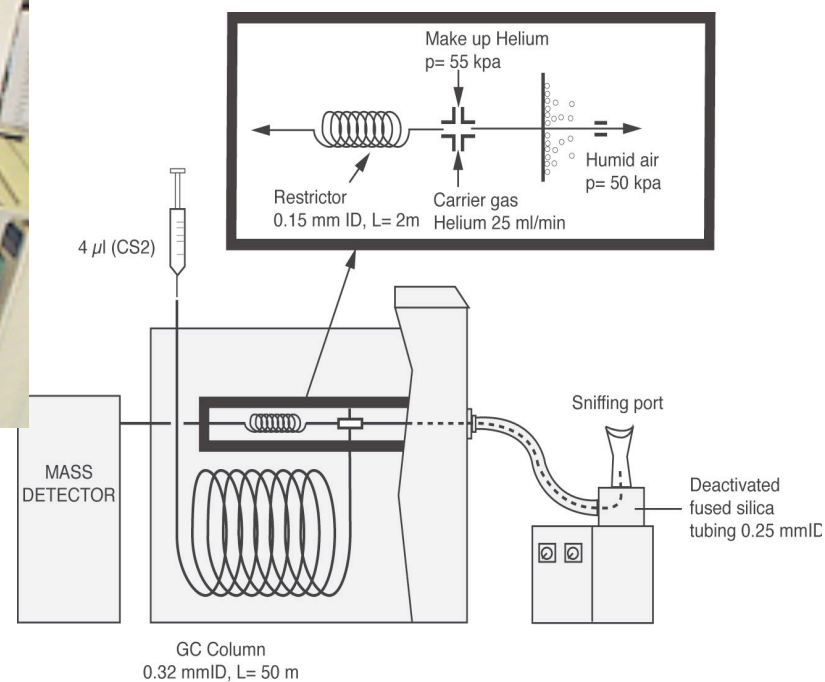
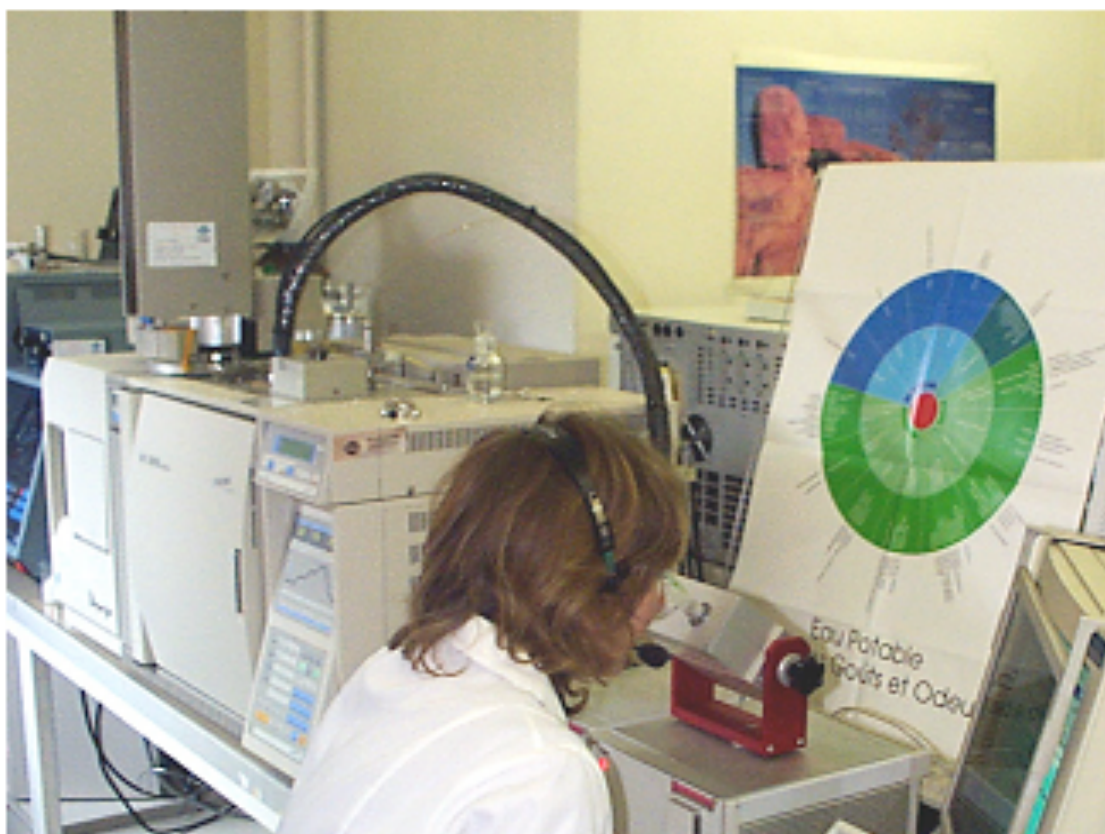
*-Also
present in
foodstuffs.*

Mass spectrum of carbonyl sulfide (O=C=S)

CONFIRMATORY ANALYSIS



CONFIRMATION OF ODORANTS BY OLFACTORY-GC/MS



Conclusions

- The **human nose** continues to be the most sensitive and quickest way to measure odor.
- The “**odor wheel**” represents an excellent tool to represent the character of the odor
- The “**odor intensity**” is found from the W-F curve
- The **Odor Profile Method** determines the Odor Character and Intensity by an Odor Panel
- **Chemical analysis** is used to backstop OPM.